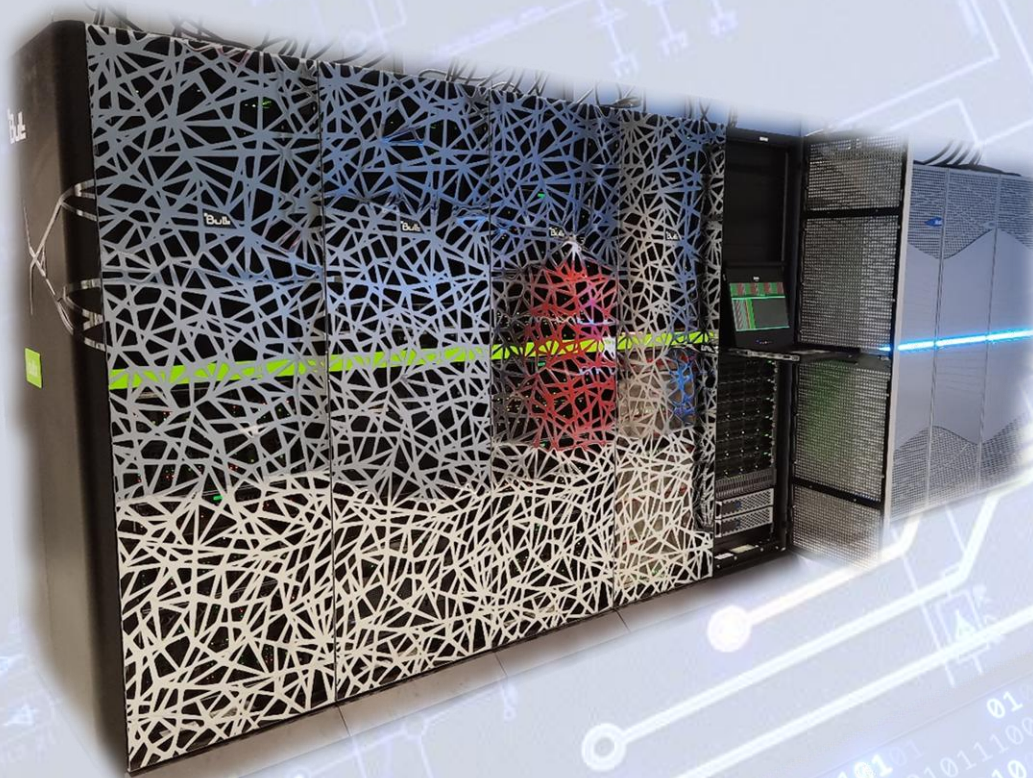




4TH EUROCC VILNIUS HACKATHON & WORKSHOP ON USING HPC



Abstract book

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Hackathon & Workshop organizers

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Scientific committee

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EuroHPC
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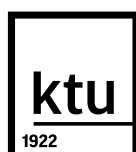
Project Implementers



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Structural characterization and accurate ^1H NMR spectra modelling of $[\text{C}_4\text{mim}][\text{BF}_4]$ ionic liquid and its aqueous mixtures

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Accurate prediction of spectroscopic features of ionic liquids is challenging because of its nanosegregated intermolecular structure, especially in their mixtures with water. Our previous work [1] showed that NMR calculations of C_4mim^+ cation ionic liquids tend to underestimate chemical shift of protons at 2nd position (H2, see Fig. 1 a). We report first accurate prediction of C_4mim^+ NMR spectra in $[\text{C}_4\text{mim}][\text{BF}_4]$ ionic liquid (IL) and water mixtures (see Fig. 1 b), including the characterization of intermolecular structure of such systems, especially the local structure around C_4mim^+ using radial and angular (Fig. 1 c) distributions. Results were achieved performing molecular dynamics (MD) simulations on four $[\text{C}_4\text{mim}][\text{BF}_4]:\text{H}_2\text{O}$ mixtures, ranging from 17 % to 100 % IL molar fraction (χ_{IL}), and then extracting 100 configurations from MD trajectories and conducting quantum mechanics / molecular mechanics (QM/MM) calculations for ^1H NMR shielding constant evaluation.

We would like to thank to "HPC Sauletekis" for providing us computational resources and Research Council of Lithuania for funding (project no. S-MIP-22-74).

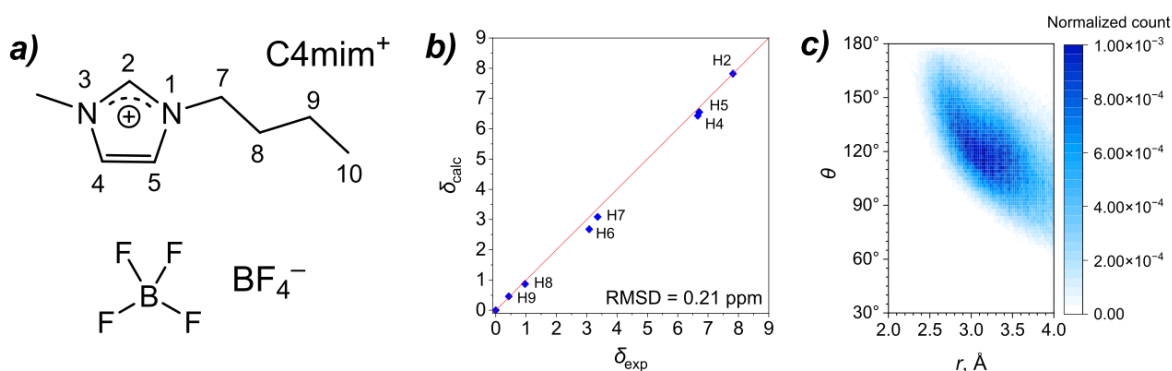


Fig. 1. $[\text{C}_4\text{mim}][\text{BF}_4]$ chemical structure (a), comparison of calculated and experimental ^1H NMR chemical shifts of C_4mim^+ cation relative to H10 protons in the butyl chain in $\chi_{\text{IL}} = 50\%$ system (b), angular distribution of BF_4^- anions around C2–H2 bond of C_4mim^+ cation in $\chi_{\text{IL}} = 50\%$ system (c).

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[1] D. Lengvinaitė, S. Kvederaviciute, S. Bielskutė, V. Klimavičius, V. Balevicius, F. Mocci, A. Laaksonen, K. Aidas; *J. Phys. Chem. B* **125** (2021) pp. 13255-13266.